Total Pages: 3

BT-I/D-19

31037

CHEMISTRY

Paper: BS-101-A

Time: Three Hours]

[Maximum Marks: 7.

Note: Attempt five questions in all selecting at least one question from each unit. All questions carry equal marks.

UNIT

- (a) Write salient features of Molecular orbital theory. Draw the molecular orbital energy level diagram of O₂ molecule, and find its bond order.
 - (b) Explain the bond structure of solids on the basis of molecular orbital theory and also explain the concept of doping.
 - (a) What do you mean by Crystal field splitting? Explain
 the splitting in octahedral field. On the basis of crystal
 field splitting explain the structure of [Co(NH₃)₆]³⁺.

(b) Why conjugated butadiene is more stake than non-conjugated butadiene.

UNIT-II

3. (a) What is Molecular spectroscopy? How does it differ from Atomic spectroscopy?

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- (b) Explain the concept of Fluorescence and how it difters from Phosphorescence. Give important applications of Fluorescence in Medicine.
- (c) Write a short note on Diffraction. 6
- (a) What do you mean by Electromagnetic spectrum? Give basic principle of Electronic spectroscopy and also explain Frank-Condon principle.
 - (b) "IR spectra is often characterised as molecular fingerprints." Justify the statement.
 5

UNIT-III

- 5. (a) Write down the important applications of Nernst equation and also derive Nernst equation. 6
 - (b) Differentiate between Gibb's free energy and Helmholz free energy, which out of these two is called Work function, and why?
 - (c) Draw molecular geometries of H₂O, PCl₅ on the basis
 of Valence shell electron pair repulsion theory.
- 6. (a) What is Concept of Entropy? Find out the expression for entropy change when volume and temperature are the two variables.
 - (b) What do you mean by Electronic configuration of element? Explain the three principles used while writing electronic configuration.

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(c) Write about Fajan rules.

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UNIT-IV

7.	(a)	What are Structural isomers? Write about position	and
		chain isomers in detail giving examples.	6

- (b) Write a note on S_{N^1} and S_{N^2} reactions with examples.
 - 6

- (c) Write Markownikof's rule with example.
- 8. (a) What are ring opening type of organic reactions?

 Explain.

 5
 - (b) Write a short note on Diastoreomers. 5
 - (c) Write a method for the synthesis of Paracetamol. Where this drug is used 5