Roll No.

SECOND SEMESTER

B.Tech. (Gp B)

END SEMESTER EXAMINATION

May-2019

AC-102 CHEMISTRY

Time: 3 Hours

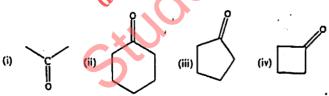
Max. Marks: 40

Note: Answer FIVE questions. Question number ONE is compulsory Assume suitable missing data, if any.

Answer the following questions (Any Six)

[2x6=12]

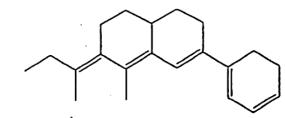
- [a] Give one example each of internal, external, self and polymeric indicator.
- [b] How DSC is used to check the surity of any compound?
- [c] Arrange the following molecules in their increasing order of stretching frequency of C=O. Discuss the reason



- [d] Write four examples of secondary batteries?
- [e] What is reduced phase rule? When is it applied?
- [f] Calculate the % atom economy of synthesis of ammonia.

$$N_2 + 3H_2 \longrightarrow 2NH_3$$

- [g] What is an auxochrome? Explain with an example.
- 2 [a] Explain the principle of DTA with the help of a suitable thermogram.
 - [b] Discuss the mode of termination in free-radical addition polymerization. Write monomers of Terylene and Kevlar. [4]
- 3 [a] Explain the mass spectrum of isopropane. Calculate λ_{max} for the following molecule using Woodward Fieser rule



[3]

- [b] What do you mean by enantiotropy? Draw and explain the phase diagram of sulphur. [4]
- 4 [a] In a polymer, there are 100 molecules of molecular weight 100; 200 molecules of molecular weight 1000; and 300 molecules of molecular weight 10,000. Find M_n, M_w and PDI.
 - [b] Predict the number of signals and multiplicity (splitting) of the respective signals in the following compounds:

- 5 [a] Explain Volhard's method for determination of halide ion. [3]
 - [b] Define electroplating. What is its significance? How the balanced concentration of metal ions is maintained in electrolytic cell? [4]
 - 6 Write short notes on any TWO of the following. [3.5x2=7]
 - (a) Li-ion battery
 - (b) Green solvents
 - (c). Fuel cell
 - (d) Eutectic system