Explain various factors that effect the relative reactivities of aldehydes and ketones.



- How can following reactions be used to prepare benzaldehyde:
 - Rosenmund's reduction
 - Etards reaction
 - (iii) Gatterman Koch aldehyde synthesis
 - (iv) Riemer Tiemann Reaction.
- Explain the following reactions:
 - Wittig reaction
 - Clemmensen reaction.

B.Sc. 4th Semester (Hons.) (New Scheme)

Examination, May-2016

CHEMISTRY

Paper-CH(H)-208 P-25

Organic Chemistry

Time allowed: 3 hours]

[Maximum marks: 40]

Note: Question No. 1 is compulsory. Select one question from each section. Attempt five questions in all. All questions carry equal marks.

- What is coupling reaction?

 - Define Fermi Resonance?
 - Why is diazotisation carried always below 278 K.
 - What is Hinsberg's reagent?
 - (e) Write chemical equation for the formation of picrates.
 - Complete the following reaction: C_cH_cCHO+Conc. NaOH →
 - Arrange the following in increasing order of their boiling point CH, CHO, CH, COCH, CH, CH, CH, CH,
 - Write any one method of preparation of Ethanenitrib.

Section-A

- Write short note on following:
 - Selection rule
 - Finger Print Region.

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	395.1 400 WCD 9334.5 WCD 9334.5 WCD 9334.5			
(b)	How will you distinguish the following pairs of the basis of IR spectroscopy.	n 5. 1	(a)	Explain any one method for the separation of mixture of 1°, 2° and 3° amines.
	(i) CH ₃ CH ₂ COOH and CH ₄ COOCH ₃		(b)	Explain the following terms:
	(ii) C ₂ H ₂ OH and C ₂ H ₂ OH			(i) Hofmann Bromamide reaction
(a)	How can IR spectroscopy be used to distinguish	h		(ii) Reduction of nitriles.
(4)	between:	1		Section-C
2	(i) Intermolecular and Intramolecular H-bonding	r 6.	(a)	Suggest a suitable mechanism for diazotisation reaction.
٠,	(ii) A phenol and an alcohol	191	(b)	Write any two method of preparation of:
(b)	Discuss briefly the principle and applications of	£		(i) Isocyanides
Δ-7	Infrared Spectroscopy.	ļ		(ii) Cyanides
	Section-B	7.	(a)	Nature of reduction product of Nitrobenzen
(a)	Explain the effect of substituents on the basicity	y		depends upon nature of reducing agent and pH or reaction medium. Explain.
	of aromatic amines.	2	(b)	How will you prepare:
(b)	Starting from aniline, how will you prepare the	e		(i) Nitroethane from Ethylchloride
	following:	2		(ii) Picric acid from Phenol
	(i) p-Bromoaniline			(iii) Nitrobenzene from benzene
1	(ii) Sulphanilic acid.			(iv) Azoxybenzene from Nitrobenzene
(c)	Write short note on the following:	ļ		Section-D
	(i) Phase transfer catalyst	8.	(a)	Explain the following reactions:
	(ii) Reductive amination of ald ehydes.		-	(i) Baeyer Villiger Oxidation of ketones
				(ii) Maranich Reaction.
2702				2