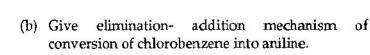
#### SECTION - IV

**8.** (a) Why aryl halides are less reactive than alkyl halides towards nucleophillic substitution reaction?



- (c) Explain the following reactions:
  - (i) Wurtz fittig reaction.
  - (ii) Ullmann biaryl synthesis.

3, 3, 2

- **9.** (a) How can you prepare (BHC) Benzene Hexachloride from benzene?
  - (b) Complete the following reactions:

(i) 
$$\bigcirc$$
 + KOH  $\longrightarrow$  (ii)  $\bigcirc$  + Na  $\longrightarrow$ 

- (c)  $CHF_3$  is less acidic than  $CHCI_3$ . Explain.
- (d) Give the mechanism of  $S_N^2$  reactions of alkyl halides. 2, 2, 2

91535- -(P-4)(Q-9)(16) (4)

Roll No.

# 91535

# B. Sc. 2nd Semester Chemistry (Hons.) (New Scheme) Examination – May, 2016

# **ORGANIC CHEMISTRY**

Paper: 203

Time: Three Hours]

[Maximum Marks: 40

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

Note: Attempt five questions in all, selecting one question from each Section. Question No. 1 is compulsory.

All questions carry equal marks.

1. (a) What is octane number?

- $1 \times 8 = 8$
- (b) What are annulenes? Give two examples.
- (c) Give Baeyer's test for Alkene.
- (d) Give the IUPAC name of DDT. Also draw its structure.
- (e) Drawmolecular orbitalstructure of Benzene.

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- (f) What is Freon? What are its uses?
- (g) What is Inversion of configuration during  $S_N^2$ ?
- (h) Name the compound used as Fire-extinguisher under the name pyrene?

## SECTION -I

- (a) Discuss the mechanism of dehydrohalogenation of alkyl halide.
  - (b) An alkene on ozonolysis give two molecules of propanone. Identify the given structure and write equation involved.
  - (c) Why the boiling point of neopentane is lower than n-pentane?
  - (d) Discuss the substitution of allylic and vinylic positions of alkene by taking suitable examples?

2, 2, 2, 2

3. (a) Complete the following reactions:

$$C_2H_5CH = CH_2 \cdot \frac{C_3}{Z_{11} \text{ dust}}$$

$$CH_3 - CH = CH_2 \frac{HBr}{\text{Peroxide}}$$

Propene + 
$$B_2H_6 \xrightarrow{H_2O_2/OH}$$

- (b) Explain the mechanism of Markownikoff's addition to alkene with statable example.
- (c) What is knocking value? What antiknock additives can be added to petrol? 3, 3, 2

91535- -(P-4)(Q-9)(16) (2)

## SECTION - II

- 4. (a) Explain the synthesis of:
  - (i) Naphthalene from benzene
  - (ii) Anthracene from buta 1,3 diene.
  - (b) Give mechanism of Electrophillic Substitution reaction in Naphthalene? 5, 3
- 5. (a) Why alkynes are acidic in nature?
  - (b) Complete the following reactions:

$$HC = CH + H_2O \xrightarrow{H_2SO_4}$$

$$HC = CH \xrightarrow{\text{Iron tube}}$$

- (c) Write a short note on Diel's Alder reaction.
- (d) Discuss the conformations of cyclohexene.

2, 2, 2, 2

# SECTION - III

- **6.** (a) What is Hackle's rule of aromaticity? Illustrate with example.
  - (b) Cycloheptatrienyl cation is more stable than cycloheptatriene. Why?
  - (c) Justify the statement that it is easy to nitrate toluene than benzene.
  - (d) Give two methods of formation of phenyl acetylene. 2, 2, 2, 2
- 7. (a) Explain the mechanism of sulphonation of benzene.
  - (b) Write a short note on Birch Reduction.
  - (c) What are antiarormatic and non-aromatic compounds? Give examples. 3, 3, 2

(3)

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P.T.O.