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- (b) Discuss the effects of nano-particle size on Raman peaks. 4

7. (a) List the factors responsible for broadening of X-ray diffraction peaks of nano materials. Discuss how this finite width of X-ray diffraction peaks is used for determination of particle size and strain in nano-materials. 11

- (b) Discuss the effect of nano-particle size on photoluminescence peak at a fix temperature. 5

#### Unit-IV

8. (a) Describe cluster beam evaporation technique for synthesis of nano-materials. Also give the list of materials deposited by this technique and its drawbacks. 9

- (b) Discuss ball milling technique for synthesis of nano-particles. How contamination from milling tools and atmospheric gases is minimized? 7

9. (a) Draw schematic diagram of chemical bath deposition technique. Explain the deposition mechanism by taking example of any particular material. Also mention its merits and demerits. 8

- (b) Describe ion beam deposition technique along with its advantages and disadvantages. 8

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M.Sc. 4th Semester (New) Examination,  
May-2016

PHYSICS

Paper-XVI

Physics of Nano-materials (New)

Time allowed : 3 hours ]

[ Maximum marks : 80

*Note : Attempt five questions in all, selecting at least one question from each unit. Question No. 1 is compulsory.*

1. (a) Discuss how the density of states varies with crystal size. Show that for simple cubic crystal density of states is proportional to cube of the lattice constant (a). 4
- (b) Explain why absorption & emission does not take place in  $Al_{0.3}Ga_{0.7}As/GaAs/Al_{0.3}Ga_{0.7}As$  quantum well structure at  $h\nu = E_g (GaAs)$ , although it happens in bulk GaAs. 4
- (c) What is Raman Effect? Why Stokes emission is much favoured over anti-Stokes at relatively lower temperatures? 4
- (d) Describe the cluster formation in context of cluster beam evaporation technique and mention the conditions under which optimum size clusters are formed. 4

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## Unit-I

2. (a) Experimentally, it is proven fact that band-gap of a semiconductor nano-crystal increases with size reduction. Prove this experimental fact theoretically i.e.

$$E_{g,nano} = E_{g,bulk} + \frac{h^2}{2} \left( \frac{1}{m_{e,eff}} + \frac{1}{m_{h,eff}} \right) \left( \frac{\pi}{R} \right)^2$$

Where R is the radius of the nano-crystal and other symbols have their usual meanings. 10

- (b) Explain the metallic character shown by divalent metals like Be, Ca, etc. Which are insulators as per predictions of the one dimensional band theory. 6
3. (a) Derive the expression for density of states for free electron i.e. three dimensional case. Plot and explain density of states versus energy graph in an energy band. 11
- (b) Give qualitative idea and features of free electron theory. 5

## Unit-II

4. (a) What do you mean by quantum confined structures and quantum confinement limit ? How many dimensions are confined and free in bulk, quantum well, wire and dot respectively ? Also draw the density of states versus energy graph for each one. 7
- (b) Write and solve Schrödinger equation for an electron confined in one and two dimensional infinite square well potential. 9
5. Discuss how a practical quantum well structure is formed. Write the condition under which this quantum well is treated as a finite well or an infinite well for an electron inside it. Solve the Schrödinger equation for the electron under infinite well approximation. Also plot its E-k diagram and density of states. 16

## Unit-III

6. (a) Describe the basic principal and instrumentation involved in photoluminescence spectroscopy. Discuss how photoluminescence spectroscopy is used in determination of alloy composition in compound semiconductors ( $Al_xGa_{1-x}As$  etc.) and in finding the well width of  $Al_xGa_{1-x}As/GsAs/Al_xGa_{1-x}As$  quantum well structure. 12