Seat No.:	Enrolment No.
See: 118:	2.0.00.000.000.000

Subject code: 110001

GUJARAT TECHNOLOGICAL UNIVERSITY B. E. - SEMESTER - I • EXAMINATION - WINTER • 2014

Date: 02-01-2015

Su	bject	Name: Chemistry	
Ti	me: 10	0:30 am - 01:00 pm Total Marks: 70	
Ins	struct	ions:	
	1. 2. 3.	Make suitable assumptions wherever necessary.	
Q.1	(a)	i) Explain the terms with examples: Lubricant, Photochemical smogii) Differentiate between chemical corrosion & electro chemical corrosion	04 03
	(b)	What is hardness of water? Calculate the temporary and total Hardness of water sample containing following data in mg/L. Mg(HCO ₃) ₂ =248; Ca(HCO ₃) ₂ =81; MgCl ₂ =190; CaSO ₄ =78 Given:molecular weight of. Mg (HCO ₃) ₂ =146; Ca (HCO ₃) ₂ =162; MgCl ₂ =95; CaSO ₄ =136.	07
Q.2	(a) (b)	What are the characteristic of a good fuel? Explain any one method of analysis. What is Galvanic Corrosion? Explain it with mechanism.	07 07
Q.3	(a)	What are the constituents of Portland cement? Discuss the mechanism of setting and hardening of Portland cement.	07
	(b)	What is galvanic corrosion? Explain its mechanism.	07
Q.4	(a) (b)	Explain the factors affecting on fermentation process in detail. Discuss:	07
	(-)	 Melt and Wet spirning Acid rain and Green house effect. 	04 03
Q.5	(a)	Explain the following term and their effects on the environment: 1. Green house effect. 2. Ozona depletion 3. Acturain.	07
	(b)	What is meant by fermentation? Discuss the manufacture of ethanol from molasses by fermentation process.	07
Q.6	(a)	Explain the terms: Paints and varnishes. Discuss types of inorganic surface coatings.	07
	(b)	i) Differentiate Renewable and Non-renewable source of energy.ii) justify the statement "Solar energy is the ultimate sources of all types of energy"	03 04
Q.7	(a) (b)	Explain the terms: Specific gravity, Melting point, Boiling point. Give Classification, properties and uses of abrasives.	07 07
