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No. of Printed Pages—4

ME-301

B. TECH.

THIRD SEMESTER EXAMINATION, 2003-2004

MATERIAL SCIENCE

Time : 3 Hours

Total Marks : 100

Note : (1) Attempt **All** six questions. There may be internal choices within questions.

- (2) Q.1 carries 20 marks and other questions carry 16 marks each. Marks are also indicated therein.
- (3) Answer briefly and neatly. No negative marking.
- (4) Symbols have their usual meaning. Assume missing data, if any.

1. Answer ALL parts as directed :— (2×10=20)

(a) Name as asked for

(i) The source energy of atom-bomb is a nuclear-process called nuclear-‘fission’. The source of solar energy in Sun is also a nuclear-process; name this process.

(ii) Einstein in 1905 gave a theory of relativity called ‘special theory of relativity’, which talks about mass increase, length contraction and time dilation and $E = mc^2$ etc. He in 1913 gave another theory of relativity which talks about that gravity is there because of curvature of space; what (name) this new theory of relativity is called ?

(b) Write material for

- (i) Engine cylinder block
- (ii) Fire-proof clothings.

- (c) Write application / use of —
- (i) Zn (chemical element)
 - (ii) PVC (plastic).
- (d) Write typical composition of
- (i) Stainless Steel
 - (ii) Gun Metal.
- (e) Write full form of
- (i) LASER (beam)
 - (ii) HSS (tool material).
- (f) Find the shortest wavelength produced in radiation for X-ray machine whose accelerating potential is 5 kilovolts.
- (g) Find the kinetic energy ($KE = mc^2 - m_0c^2$) of an electron (in eV) moving with half the velocity of light. Rest mass (energy) of electron in electron volts is 0.51 eV.
- (h) Copper (FCC) has atomic weight 63.5 and atomic radius 1.78 Å. Find its density in gram/cc.
- (i) X-ray with wavelength 0.58 Å is used for evaluating interplanar spacing between certain planes. First order diffraction is obtained at glancing angle 9.5° . Find the interplanar spacing d . Also find angle for 3rd order diffraction.
- (j) A bolt has to take up 1 ton (1000 kgf) tensile load. The yield strength of bolt-material is 400 N/mm². Design the diameter of bolt, taking a factor of safety as 2 i.e., permissible stress is limited to only half the yield stress.

2. Answer any TWO of the following :— (8×2=16)

- (a) Briefly describe Radioactivity and properties of α , β and γ rays. Also derive $t_{1/2} = 0.69 / \lambda$.
- (b) Describe (enlist / classify) various types of Bravais lattices for crystals.
- (c) Enlist and briefly describe various types of imperfections (defects and dislocations) in solids.

Answer any TWO of the following :— (8×2=16)

- (a) Briefly explain (i) Fatigue and (ii) Creep and its testings. Also mention situations where these are important.
- (b) Draw σ - ϵ diagrams and microstructure of (i) mild steel and (ii) gray cast-iron, and also mention % C in these.
- (c) Draw neat sketch of Fe-C equilibrium diagram and briefly describe cooling of 0.3 % carbon steel from molten stage to room temperature.

4. Answer any TWO of the following :— (8×2=16)

- (a) Briefly describe any 2 steel making furnaces. Also name the furnace used for making wrought-iron and cast-iron. Differentiate between Carbon-steels and Alloy-steels.
- (b) Briefly describe Annealing, Normalising, Quenching and Tempering heat-treatment processes and show these processes on TTT diagram.
- (c) Differentiate between Brass and Bronze. Also differentiate between Soft-brass and Hard-brass and mention their applications.

5. Answer any TWO of the following :— (8×2=16)

- (a) Differentiate between Soft and Hard magnetic materials. Also briefly describe the principle of magnetic-storage.
- (b) Briefly describe solid-diffusion and factors affecting it. Write applications of solid-diffusion, specially in reference to doping of semi-conductors and its devices.
- (c) Differentiate between Type I and Type II superconductors. Mention the applications of superconductors. Write the full form of : BCS (theory), JJ, SQUIDS related to superconductors.

6. Answer any TWO of the following :— (8×2=16)

- (a) Briefly describe types and applications of Ceramics. Also show that cation to anion radius ratio is 0.155 for co-ordination number 3 (i.e., a small cation is compactly surrounded by 3 big anions).
- (b) Enlist five Plastics and its chemistry (formula) and applications. Also briefly describe injection-moulding of plastic.
- (c) Briefly describe types of Corrosion and Methods for corrosion-prevention.